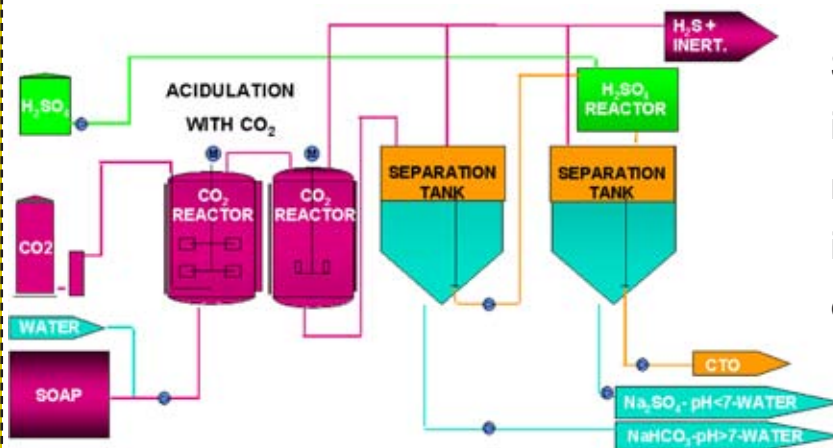


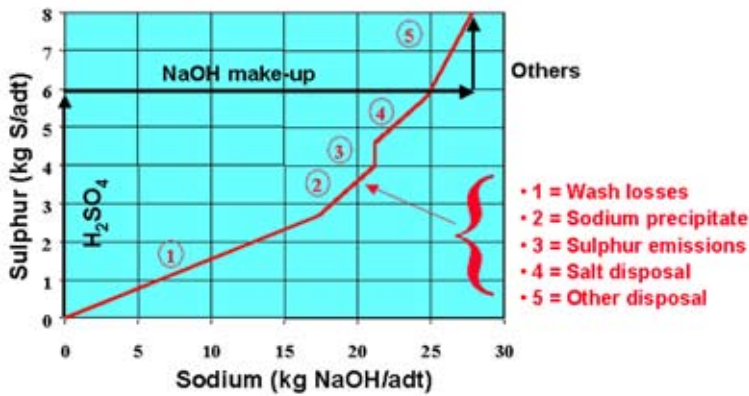
PULP & PAPER



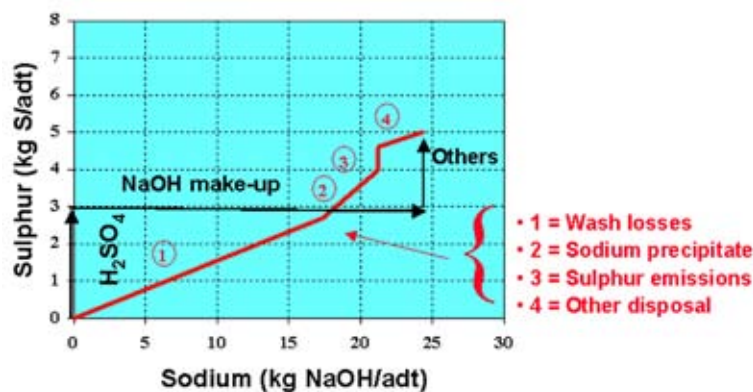
CO₂ for Tall Oil recovery



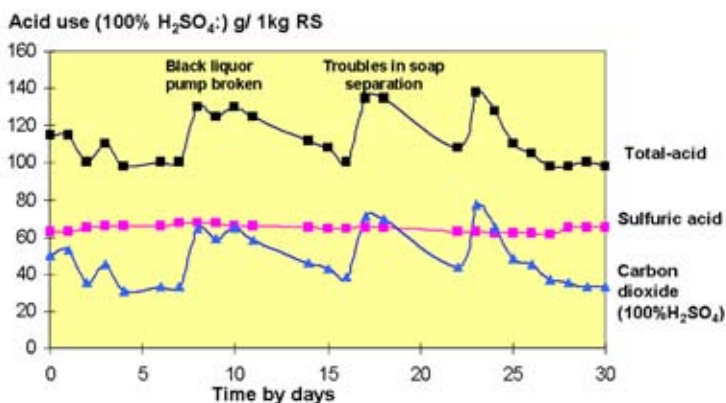
Several pulp mills have successfully implemented this process which has resulted in greater savings and has improved the environmental impact of Tall Oil recovery.



Sodium-sulphur balance without CO₂



Sodium-sulphur balance with CO₂



Trend of acid use in soap oil cooking

Replacing H₂SO₄ with CO₂ for Tall Oil recovery provides significant benefits:

- it lowers sulphur input in the recovery cycle and allows better control of the Na/S balance
- it reduces the consumption of H₂SO₄ and NaOH
- it improves Tall Oil quality
- it lowers sulphur emissions in air and water

Tall Oil recovery with CO₂

Acidification of the Tall Oil soap recovered from the black liquor is performed in two steps:

- a neutralization step with gaseous CO₂ followed by a proprietary treatment and a separation stage (the NaHCO₃ brine is sent to recovery)
- a final acidification step with H₂SO₄ (merchant or recovered from ClO₂ generator) and a separation stage (the Na₂SO₄ brine is sent to recovery)

Facts and Figures

The first CO₂ Tall Oil Recovery unit was installed in 1994 at the Metsä-Botnia mill in Kemi, Finland, and has been running smoothly ever since.

Roughly 50% of the previously used H₂SO₄ has been replaced by CO₂ (70-120 kg CO₂ per ton of Tall Oil). As a result of this lower sulphur input in the Kraft cycle (see figures), improvements have been realized in terms of process operation, economics and environmental impact

- Significant savings of NaOH make-up thanks to lower ash salt disposal from precipitator (see figures)
- Possible use of waste H₂SO₄ from ClO₂ generators instead of merchant H₂SO₄
- Recovery cycle easier to control and run
- Less malodorous emissions of sulphurous gases

The Kemi mill has also found an improvement of Tall Oil quality and yield. This patent pending process which combines economic efficiency and environmental soundness is provided by Air Liquide on a worldwide basis.



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